A FACILE ROUTE TO CYCLOPENTENONES BY FRIEDEL-CRAFTS ACYLATION

Salih HACINI[†]' Roselyne PARDO and Maurice SANTELLI*

(Centre de St-Jérome, Rue H. Poincaré, 13397-MARSEILLE Cedex 4 France)

<u>Summary</u>: Various cyclopentenones are obtained in fair to high yields when ethylenic acyl bromides are condensed with cyclohexene at low temperature. With less reactive reagents, condensation occurs at 0 °C, leading to β -chloroketones. A muscone precursor is obtained in two steps with excellent yields.

Cyclopentenones are useful intermediates in the synthesis of a wide range of compounds as prostaglandins (1), muscone (2), terreins (3), antitumoral substances (4), rethrolones (5) etc.

Usually, cyclopentenones have been produced in good yields by intramolecular aldol ring closures of 1,4-diones (6). The route to cyclopentenones <u>via</u> cyclisation of divinylketones (NAZAROV-type conrotatory electrocyclisation (7) is an attractive one, limited only by the availability of the divinylketones (generally prepared by a RUPE rearrangement of acetylenic diols (2,8), acylation of alkenes with unsaturated acids in polyphosphoric acide (9) or acylation of vinylsilanes (10)).One another route can be found in the FRIEDEL-CRAFTS acylation of alkenes by ethylenic acid halides in the presence of AlCl₃ (earlier studies had shown that polyhalogenated ketones are produced (11)).



<u>Results</u>: In a preliminary experiment, condensation of cyclohexene with crotonoyl chloride, at 0 °C, gave a mixture, the main component of which is trans- β -chloroketone $\underline{3}$ (12) :



[†]On leave from the University of Oran (Algeria)

Chloroketone $\underline{3}$ is practically the only primary product, ketones $\underline{1}$ (13) and $\underline{2}$ (9a) resulting from the work-up.

Nucleophilic reagents, such as complex hydrides, pyrrolidine or trimethylallylsilane (14), have been found to add very easily to the dienone $\underline{\underline{1}}$ on the propenyl moiety :



For this reason, we conclude that the chloroketone $\underline{3}$ is not produced by conjugate addition of HCl on the dienone $\underline{1}$. Accordingly, we were led to consider the β -ketocarbenium ion $\underline{9}$ as precursor of chloroketone $\underline{3}$ and dienone $\underline{1}$:



These were reasons to expect that the importance of the competing H^+ abstraction reaction could be increased by lowering the reaction temperature. In fact, by using the more reactive crotonoyl bromide (15), condensation occurs at low temperature (-78 °C) and bicyclic ketone $\underline{2}$ was isolated in an 80 % yield (16):



The present reaction is useful especially for the preparation of 6/5 fused bicyclic ketones in fair to excellent yields:



.

When the same reaction conditions were applied to the trisubstituted alkenes, the main product is in fact often the allylvinylketone (17):



Condensation of cyclohexene with sebacoyl bromide at low temperature does not occur, and warming up to 0 °C (during 18 h) is necessary; chloro-ketone 21 is, as expected, the major product:



In the same way, cyclododecene (cis + trans) reacts slowly with crotonoyl bromide at -78 $^\circ\text{C}$:



Ketone $\underline{23}$ is an important intermediate in the synthesis of (-)-muscone (2). By acid treatment (HCOOH-H₃PO₄ 3/1 (w/w), 80 °C, 100 h), the chloroketone $\underline{22}$ is converted easily to the cyclopentenone $\underline{23}$ (quantitative yield). This method has the advantage of a direct and **simple** synthesis using very cheap reagents, in contrast to previous more sophisticated methods (2, 10, 18)

It is clear that the FRIEDEL-CRAFTS acylation with unsaturated acyl bromides has still a large scope in synthesis.

References and Notes :

- (1) (a) A. MITRA, "The Synthesis of Prostaglandins", Wiley, New-York, 1977.
 (b) J.S. BINDRA and R. BINDRA, "Prostaglandin Synthesis", Academic Press,
 - New-York, 1977.
 (c) "Prostaglandin Research", Ed. P. CRABBE, Academic Press, New-York, 1977.

- (2) (a) M. BAUMANN, W. HOFFMANN and N. MULLER, Tetrahedron Letters, 1976, 3585.
 (b) M. KARPH and A.S. DREIDING, Helv. Chim. Acta, 1976, <u>59</u>, 1226.
 (c) R.W. GRAY and A.S. DREIDING, ibid., 1977, <u>60</u>, 1969.
- (3) (a) J. AUERBACH and S.M. WEINREB, J. Chem. Soc., Chem. Comm., 1974, 298.
 (b) D.H.R. BARTON and L.A. HULSHOP, J. Chem. Soc. Perkin I, 1977, 1103.
- (4) K.H. LEE, E.C. MAR, M. OKAMOTO and I.H. HALL, J. Med. Chem., 1978, 21, 819.
- (5) "General and Synthetic Methods", The Chemical Society, London, 1979, p.303
- (6) R.A. ELLISON, Synthesis, 1973, 397. For other examples of the 1,4-dione cyclisation, see :
 G. STORK and M.E. JUNG, J. Amer. Chem. Soc., 1974, 96, 3682;
 M. MIYASHITA, T. YANAMI and A. YOSHIKOSHI, ibid., 1976, <u>98</u>, 4679;
 E. KEINAN and Y. MAZUR, ibid., 1977, <u>99</u>, 3861;
 T. CUVIGNY, M. LARCHEVEQUE and H. NORMANT, Tetrahedron Letters, 1974, 1237;
 E. WADA, T. NAKAI and M. OKAWARA, Chem. Letters, 1976, 1025;
 P.A. GRIECO and Y. OHFUNE, J. Org. Chem., 1978, <u>43</u>, 2720.
- (7) (a) I.N. NAZAROV and I.I. ZARETSKAYA, Zh. Obsh. Chem., 1957, <u>27</u>, 693.
 (b) C.W. SHOPPEE and B.J.A. COOKE, J. Chem. Soc. Perkin I, 1972, 2271 and 1973, 1026
 - (c) R.B. WOODWARD and R. HOFFMANN, "The Conservation of Orbital Symmetry", Verlag Chemie, Weinheim. 1970, p. 58.
 - (d) T.S. SORENSEN and A. RAUK, "Pericyclic Reactions", Vol. 2, A.P.MARCHAND and R.E. LEHR, Eds. Academic Press, New-York, 1977, p. 31.
- (8) (a) A.M. ISLAM and R.A. RAPHAEL, J. Chem. Soc., 1953, 2247.
 G.A. MACALPINE, R.A. RAPHAEL, A. SHAW, A.W. TAYLOR and H.J. WILD, J. Chem. Soc., Perkin Trans. I, 1976, 410.
 - (b) T. HIYAMA, M. SHINODA and H. NOZAKI, J. Amer. Chem. Soc., 1979, 101, 1599.
- (9) (a) S. DEV, J. Indian Chem. Soc., 1957, 34, 169.
 C. RAI and S. DEV, ibid., 1957, 34, 177.
 S.B. KULKARNI and S. DEV, Tetrahedron, 1968, 24, 545.
 - (b) L. RAND and R.J. DOLINSKI, J. Org. Chem., 1966, 31, 3063.
 - (c) J.M. CONIA and M.L. LERIVEREND, Bull. Soc. Chim., 1970, 2981.
- (10) (a) F. COOKE, J. SCHWINDEMAN and P. MAGNUS, Tetrahedron Letters, 1979, 1995
 (b) W.E. FRISTAD, D.S. DIME, T.R. BAILEY and L.A. PAQUETTE, ibid., 1979, 1999.
- (11) (a) G. BADDELEY, H.T. TAYLOR and W. PICKLES, J. Chem. Soc., 1953, 124.
 (b) N. JONES and H.T. TAYLOR, ibid., 1959, 4017.
 (c) A.B. SMITH and W.C. AGOSTA, J. Amer. Chem. Soc., 1973, <u>95</u>, 1961.
- (12)For nmr studies of the stereochemistry of the acylation of cyclohexene, see: L. OTVOS, H. TUDOS and L. RADICS, Chem. & Ind., 1970, 597.
- (13)Y. KASHMAN and H. RONEN, Tetrahedron, 1973, 29, 4275.
- (14)R. PARDO, J.P. ZAHRA and M. SANTELLI, Tetrahedron Letters, following paper
- (15) J.K. GROVES, Chem. Soc. Rev., 1972, 1, 73.
- (16) In a typical experiment, anhydrous AlCl₃ (10 mmol) was stirred with CH₂Cl₂ (25 ml) and acyl bromide (10 mmol) at room temperature until dissolution. The mixture was cooled to -78 °C and alkene (20 mmol) was added. The mixture was kept at -78 °C for 30 h and was then hydrolysed (basic work-up). With less reactive alkenes, temperature was raised from -78 °C to 0°C over 24 h.
- (17)Formation of allylvinylketone has been observed but less reactive reagents have to be used, for example see:E. KLEIN and W. ROJAHN, Tetrahedron Letters, 1971, 3607.

E. KLEIN and W. KOJANN, TELTANEUTON LELLETS, 1971, 3007.

(18)M. BERTRAND, J.P. DULCERE, G. GIL and M.L. ROUMESTANT, Tetrahedron Letters 1979, 1845.

(Received in France 30 July 1979)